

(19) World Intellectual Property Organization  
International Bureau



(43) International Publication Date  
22 July 2010 (22.07.2010)

PCT

(10) International Publication Number  
**WO 2010/081749 A1**

- (51) **International Patent Classification:**  
*C08F 2/50* (2006.01)      *C08F 2/48* (2006.01)
- (21) **International Application Number:**  
PCT/EP2010/050003
- (22) **International Filing Date:**  
4 January 2010 (04.01.2010)
- (25) **Filing Language:** English
- (26) **Publication Language:** English
- (30) **Priority Data:**  
09150509.9      14 January 2009 (14.01.2009)      EP
- (71) **Applicant (for all designated States except US):** BASF SE [DE/DE]; 67056 Ludwigshafen (DE).
- (72) **Inventors; and**
- (75) **Inventors/Applicants (for US only):** ONCLIN, Steffen [NL/CH]; Engelgasse 102, CH-4052 Basel (CH). LUCIC, Svetozar [CH/CH]; Sagiweg 1, CH-5082 Kaisten (CH). POWELL, Karin [DE/DE]; Fridolinsgasse 4, 79540 Lörrach (DE). PETER, Wolfgang [DE/DE]; Murtenerstrasse 14, 79395 Neuenburg am Rhein (DE). PIRRUNG, Frank Oliver Heinrich [DE/DE]; Krumme Länge 13, 79589 Binzen (DE). AUSCHRA, Clemens [DE/DE]; Rennweg 7, 79106 Freiburg (DE).
- (74) **Common Representative:** BASF SE; IP Department, P.O. Box, CH-4002 Basel (CH).
- (81) **Designated States (unless otherwise indicated, for every kind of national protection available):** AE, AG, AL, AM, AO, AT, AU, AZ, BA, BB, BG, BH, BR, BW, BY, BZ, CA, CH, CL, CN, CO, CR, CU, CZ, DE, DK, DM, DO, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, GT, HN, HR, HU, ID, IL, IN, IS, JP, KE, KG, KM, KN, KP, KR, KZ, LA, LC, LK, LR, LS, LT, LU, LY, MA, MD, ME, MG, MK, MN, MW, MX, MY, MZ, NA, NG, NI, NO, NZ, OM, PE, PG, PH, PL, PT, RO, RS, RU, SC, SD, SE, SG, SK, SL, SM, ST, SV, SY, TH, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, ZA, ZM, ZW.
- (84) **Designated States (unless otherwise indicated, for every kind of regional protection available):** ARIPO (BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW), Eurasian (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European (AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HR, HU, IE, IS, IT, LT, LU, LV, MC, MK, MT, NL, NO, PL, PT, RO, SE, SI, SK, SM, TR), OAPI (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG).
- Published:**  
— with international search report (Art. 21(3))



WO 2010/081749 A1

(54) **Title:** LIQUID PHOTOINITIATOR BLEND

(57) **Abstract:** The present invention relates to a homogenous liquid blend comprising a photoinitiator and an alkoxyolate surfactant for incorporation into aqueous based and solvent based radiation curable systems. In particular the invention relates to a homogenous liquid blend which is liquid at room temperature comprising at least 30% by weight with regard to the whole blend of a solid photoinitiator of the radical type and 1-70% by weight of at least one alkoxyolated liquid surface active agent and optionally an organic diluent.

Liquid photoinitiator blend

The present invention relates to a homogenous liquid blend comprising a photoinitiator and an alkoxyate surfactant for incorporation into aqueous based and solvent based radiation curable systems.

5

A potential problem is that many photoinitiators are solid at room temperature and have to be dissolved before incorporating them into coating formulations. Solubilizing solid photoinitiators into the coating formulation may require excess or additional diluents that could adversely affect the properties of the coating or may be undesirable from environmental aspects. Moreover, the use of solid photoinitiators often results in processing disadvantages during manufacture of UV-curable paints. They often require longer times of mixing until a uniform distribution and dissolution has been achieved. In addition, especially with water-borne UV-curable paints, it is also very difficult to control the status of homogeneous dissolution as the paint formulation is in itself turbid. These problems lead to higher processing effort and less reproducible quality. Therefore liquid photoinitiators are preferred for producing of UV-curable paints. Other options include dissolving the photoinitiators in the reactive diluent, dissolving a solid photoinitiator in a liquid one, and dissolving the photoinitiators in nontoxic, nonvolatile solvents that do not adversely affect coating properties.

10

15

20

US 4,609,612 (Ciba) addresses the problem of mixing benzoylcyclohexanol which is a crystalline compound melting at 50°C into oily-liquid or pasty photopolymerisable compositions. The problem is solved by mixing the 1-benzoylcyclohexanol with equal parts of benzophenone which melts below 80° C. These mixtures are liquid within a certain mixing range, and can therefore be quickly incorporated in a simple manner into liquid or pasty compositions. This solution is not suitable for aqueous systems.

25

The European Patent EP1230276B1 (Ciba) relates to aqueous suspensions of bis(2,4,6-trimethylbenzoyl)-phenylphosphine oxide (IRGACURE® 819). A dispersing agent is dissolved in water and the photoinitiator is added at room temperature. The suspension obtained is then treated in a colloid mill to obtain particles in the range of 0.1-12 µm.

30

The European Patent EP386650 (Lamberti) discloses an aqueous dispersion comprising a photoinitiator and a surfactant. The photoinitiator is present from 30 to 60% by weight of the

disperse system. The surfactant is e.g. a reaction product of a higher aliphatic alcohol with ethylene oxide and propylene oxide. At first a homogeneous solution is prepared by adding the photoinitiator at elevated temperatures to the surfactant. To the homogeneous solution water is added. The solution thus cools down and solidifies. A dispersion (solid particles) of the photoinitiator is obtained. The dispersion is grinded. As an aqueous dispersion there still remain disadvantages in that the photoinitiator particles may settle on prolonged storage and that the aqueous dispersion is incompatible to many non-aqueous UV-systems.

UK Patent Application GB 2014152 describes a blend comprising a triarylsulphonium complex salt photoinitiator dissolved in an alkoxyated liquid surface active agent to form liquid solutions in which the complex salt is at a concentration of at least 5% by weight and may be as high as 30% by weight. Higher concentrations of the photoinitiator are not disclosed.

The problem underlying the present invention is to provide a liquid photoinitiator blend which remains liquid and homogeneous (i.e. no formation of a dispersion) at room temperature and which has a high active content of photoinitiator, preferably more than 30% by weight. Said blend should be suitable for aqueous based and solvent based radiation curable systems.

It has now been found that mixing a photoinitiator of the radical type with a liquid surface active agent, and optionally with an organic diluent leads to a homogenous liquid blend with a high content of photoinitiator and which is stable at room temperature.

Thus, the invention relates to a homogenous liquid blend which is liquid at room temperature comprising at least 30% by weight with regard to the whole blend of a solid photoinitiator of the radical type and 1-70% by weight of at least one alkoxyated liquid surface active agent and optionally an organic diluent. (Claim 1)

Compared to the dispersed photoinitiator of EP386650 (Lamberti) which is a heterogeneous system the present invention is a homogeneous system at room temperature. Lamberti requires water to obtain a heterogeneous product form at room temperature. Also high energy input is required to prepare the product form (dispersing process).

**Definitions**

The term "**homogenous liquid blend**" generally describes a substance in a liquid phase, without the presence of a secondary droplet or particle phase that could separate. The liquid blend has at room temperature a viscosity not higher than 30000 mPas and therefore can be processed as a liquid with standard technical equipment for pumping and mixing of liquid materials.

The **alkoxylated surface active agent** is responsible for the emulsifying properties and for providing compatibility of the resulting photoinitiator additive concentrate towards aqueous and non-aqueous formulations like paint or ink formulations. Preferably, the pure surfactant is a liquid at room temperature.

The term "alkoxylated liquid surface active agent" is understood to include different types of non-ionic compounds of surface active agents selected from alkoxylated compounds based on adducts of ethylene oxide and/or propylene oxide. (claim 3)

The term "alkoxylated liquid surface active agent" refers to compounds in which one or more straight or branched C<sub>1</sub>-C<sub>22</sub> alkyl chains are connected through the residue of a linking group to an oligomeric or polymeric chain of alkylene oxide residues. The term also refers to compounds in which one or more aromatic groups, which can be alkylated, are connected through the residue of a linking group to an oligomeric or polymeric chain of alkylene oxide residues. The term also refers to compounds containing one or more acetylenic groups, which can be connected through the residue of a linking group to one or two oligomeric or polymeric chains of alkylene oxide residues. The term also refers to block-copolymers or random copolymers of ethylene oxide and propylene oxide. The linking group can be an oxygen atom (hydroxyl group residue from alcohol or phenol); a carboxyl group (fatty acid or ester residue); an amino group (amine group residue); or a carboxyamido (carboxylic amide residue). The alkylene oxide residues are typically residues of ethylene oxide -CH<sub>2</sub>-CH<sub>2</sub>-O- or propylene oxide -CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-O- , (-CH(CH<sub>3</sub>)-CH<sub>2</sub>-O) or combinations of ethylene and propylene oxide residues.

The surface active agent is thus selected from alcohol alkoxylates (glycol ethers), fatty acid alkoxylates, fatty amine alkoxylates, or fatty amide alkoxylates, alkoxylates of phenols or alkylsubstituted phenols, block copolymers or random copolymers of ethylene oxide and

propylene oxide, alkoxyates of alkine-alcohols or alkine-diols, alkoxyates of castor oil, alkoxyates of sorbitan fatty acid esters and mixtures thereof. (Claim 4)

Examples of suitable alkoxyated surface active agents include:

- 5 alcohol alkoxyates (glycoethers), of the formula  $C_1-C_{22}alkyl-O-(AO)_n-H$  or  $C_1-C_{22}alkyl-O-(AO)_n-C_1-C_6alkyl$ ;  
 fatty acid alkoxyates of the formula  $C_1-C_{22}alkyl-COO-(AO)_n-H$  or formula  $C_1-C_{22}alkyl-COO-(AO)_n-C_1-C_6alkyl$ ;  
 fatty amine alkoxyates of the formula  $C_1-C_{22}alkyl-NR_a-(AO)_n-H$ ;  $C_1-C_{22}alkyl-NR_a-(AO)_n-C_1-C_6alkyl$ ;  
 10 or fatty amide alkoxyates of the formula  $C_1-C_{22}alkyl-CO-NR_a-(AO)_n-H$ ;  $C_1-C_{22}alkyl-CO-NR_a-(AO)_n-C_1-C_6alkyl$ ;  
 wherein  $R_a$  is  $C_1-C_6alkyl$  or a group  $-(AO)_n-H$  and each AO is independently an ethylene oxide or propylene oxide, and the total of the indices n in the molecule is preferably in the  
 15 range from 2 to 80, more preferably 3 to 40;  
 alkoxyates of phenols or alkylsubstituted phenols;  
 block copolymers or random copolymers of ethylenoxide and propylene oxide;  
 alkoxyates of alkine-alcohols or alkine-diols;  
 alkoxyates of castor oil;  
 20 alkoxyates of sorbitan fatty acid ester.

The  $C_1-C_{22}alkyl$  group in the above mentioned alkoxyates is linear or branched, saturated or unsaturated and preferably  $C_6-C_{20}$ .

- 25 Examples of suitable alkoxyated liquid surface active agents include also compounds in which a straight or branched  $C_1-C_{22}alkyl$  chain is connected through the residue of a linking group as defined above to multifunctional alcohols like trimethylolpropane, sugar derivatives like sorbitan esters, triglycerides like castor oil and the like.
- 30 Preferred are non-ionic surfactants having a poly(alkylene oxide) group, such as a poly(ethylene oxide) group or a poly(propylene oxide) group or mixtures thereof.

Examples are compounds sold under the Trade Mark Brij (such as polyoxyethylene (4) lauryl ether (Brij 30) or polyoxyethylene (20) cetyl ether (Brij 58), polyoxyethylene (20) stearyl ether

(Brij 78) and polyoxyethylene (20) oleyl ether (Brij 92) or polyethylene glycol monomethyl ether (MPEG). Other examples are adducts to C13-oxoalcohols like C13-oxoalcohol + 3 EO (Lutensol TO 3), C13-oxoalcohol + 7 EO (Lutensol TO 7), or adduct to C10-oxoalcohols like C10-oxoalcohol + 6 EO (Lutensol ON 60).

5

Also preferred are non-ionic surfactants derived from ethoxylated esters of fatty acids with glycerine or sorbitan, such as polyoxyethylene glycerine monostearate, polyoxyethylene sorbitan monolaurate, polyoxyethylene sorbitan monopalmitate, polyoxyethylene sorbitan-monostearate, polyoxyethylene sorbitan monooleate, polyoxyethylene sorbitan tristearate, polyoxyethylene sorbitan trioleate.

10

The sorbitan esters are sold under the tradename Tween, like polyoxyethylene (20) sorbitan monolaurate (Tween 20), polyoxyethylene (20) sorbitan monopalmitate (Tween 40), polyoxyethylene (20) sorbitan monostearate (Tween 60), polyoxyethylene (20) sorbitan tristearate (Tween 65), polyoxyethylene (20) sorbitan monooleate (Tween 80), polyoxyethylene (20) sorbitan trioleate (Tween 85).

15

Examples of suitable alkoxyated fatty acids are: Diethylene glycol mono-oleate (Radiesurf 7400), PEG-400 monolaurate (Radiesurf 7423), PEG-600 dioleate (Radiesurf 7444),

20

Examples of suitable fattyamine alkoxyates are adducts of laurylamin with 5 EO (Lutensol FA 5 K), adducts of laurylamin with 10 EO (Lutensol FA 10 K),

Examples of suitable alkoxyated fattyamides are adducts of oleic acidamide and 10 EO (Lutensol FSA 10) or the adduct of oleic acid monoethanolamide with 3 EO (Serdolamide NXC 3 H)

25

Examples for alkoxyated phenols are adducts of nonylphenol with 3 EO (Marlophen NP 3), adduct of nonylphenol with 10 EO (Marlophen NP 10), or alkylenoxide adducts of tristyrylphenol.

30

Examples of block copolymers of ethylenoxide and propylene oxide are the products sold under the Pluronic trademark like: Pluronic 3100, Pluronic 3500, Pluronic 4200, Pluronic 6400 or Tetronic RED 9040 (EO-PO block copolymer adduct to ethylene diamine)

Examples of alkoxyates of alkine-alcohols or alkine-diols are product sold under the Trademarks: Surfynol (like Surfynol 104, Surynol 420, Surfynol 440, Surfynol 465 and Surfynol 2502)

- 5 Also preferred are non-ionic surfactants derived from ethoxyates of vegetable oils or animal fats, such as corn oil ethoxylate, castor oil ethoxylate, tallow oil ethoxylate.

An example for a castor oil ethoxyates is one marked under the trademark Hedipin.

- 10 The surfactant can be a blend of all classes of alkoxyated surface active substances described previously.

Optionally one or more organic diluents may be present.

- 15 The optional **organic diluent** provides additional stability of the photoinitiator concentrate and assists in improving compatibility of the photoinitiator concentrate towards different end use formulations.

The optional organic diluent is chosen from the group consisting of high boiling inert liquids  
20 based on ester and or ether compounds like: oligo ethylene glycols, adipates, phthalates, alkyl esters, aromatic esters, diesters, etc. Also suitable are high boiling liquid alcohols like C10 to C20-alcohols or polyols. More preferably the organic diluent has a boiling point above 120°C, most preferred above 160°C.

Examples of oligoethylene glycols include: triethylene glycol, tetraethylene glycol or  
25 tripropylene glycol or tetrapropylene glycol or methoxypolyethylene glycols of different molecular weights. Examples of organic diluents based on ester fluids include: dioctyl adipate, dibutyladipate, benzyl benzoate, dimethyl phthalate, dibutylphthalate, fatty acid methylesters, fatty acid ethylhexylesters, citric acid triethylester.

Examples of ether fluids are diethylene glycol dibutyl ether and phenoxyethanol

30 Examples of alcohols are isodecanol, C12-C15 oxoalcohols, propylene glycol, dipropylene glycol, tripropylene glycol, oligoethylene glycols like diethylene glycol, triethylene glycol, tetraethylene glycol and glycerin

In one embodiment the organic diluent is present. (claim 5)

The amount of organic diluent in the inventive liquid blend is typically less than 60% by weight, preferably less than 40% and most preferably less than 25%.

5 Optionally the blend may contain water. Small amounts of water may not have a negative impact concerning the stability of the inventive photoinitiator blends as well as concerning the compatibility of the inventive photoinitiator blends for non-aqueous systems.

The amount of water in the inventive liquid blend is typically less than 10% by weight, preferably less than 5% and most preferable less than 2% or water-free.

10

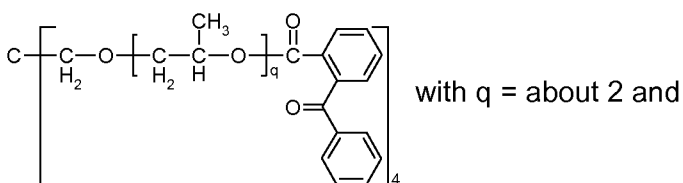
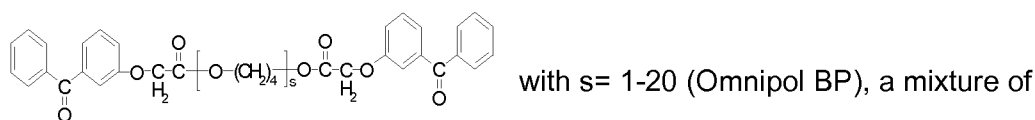
**Photoinitiators**

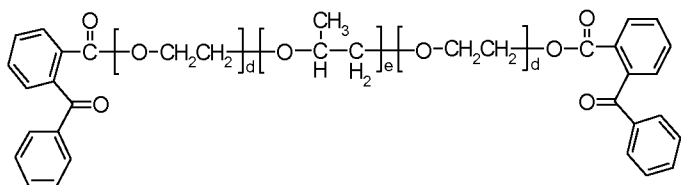
The photoinitiator is of the radical type. The photoinitiator is solid at room temperature.

In one embodiment the photoinitiator is selected from benzophenones, aromatic  $\alpha$ -hydroxyketones, benzilketals, aromatic  $\alpha$ -aminoketones, phenylglyoxalic acid esters, mono-acylphosphinioxides, bis-acylphosphinioxides, tris-acylphosphinioxides and/or oximesters derived from aromatic ketones. (claim 6)

Examples of photoinitiators are camphor quinone; benzophenone, benzophenone derivatives, such as 2,4,6-trimethylbenzophenone, 2-methylbenzophenone, 3-methylbenzophenone, 4-methylbenzophenone, 2-methoxycarbonylbenzophenone 4,4'-bis(chloromethyl)-benzophenone, 4-chlorobenzophenone, 4-phenylbenzophenone, 3,3'-dimethyl-4-methoxybenzophenone, [4-(4-methylphenylthio)phenyl]-phenylmethanone, methyl-2-benzoylbenzoate, 3-methyl-4'-phenylbenzophenone, 2,4,6-trimethyl-4'-phenylbenzophenone, 4,4'-bis(dimethylamino)benzophenone, 4,4'-bis(diethylamino)benzophenone,

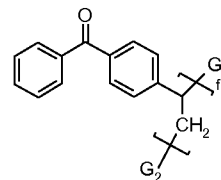
25



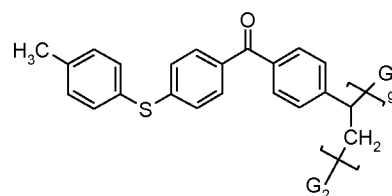


wherein the sum of d and e is about

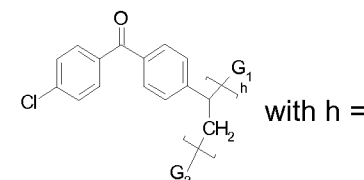
14, where d is greater than e (Speedcure® 7005, provided by Lambson),



with f = about 14 (Speedcure® 7006, provided by Lambson);



with g = about 12 (Speedcure® 7003, provided by Lambson);



with h =

- 5 about 13 (Speedcure® 7020, provided by Lambson), and any blends or admixtures of the above mentioned compounds; thioxanthenes, thioxanthone derivatives, polymeric thioxanthenes as for example OMNIPOL TX; ketal compounds, as for example benzildimethylketal (IRGACURE® 651); acetophenone, acetophenone derivatives, for example  $\alpha$ -hydroxycycloalkyl phenyl ketones or  $\alpha$ -hydroxyalkyl phenyl ketones, such as for example 2-hydroxy-2-methyl-1-phenyl-propanone (DAROCUR® 1173), 1-hydroxy-cyclohexyl-phenyl-ketone (IRGACURE® 184), 1-(4-dodecylbenzoyl)-1-hydroxy-1-methyl-ethane, 1-(4-isopropylbenzoyl)-1-hydroxy-1-methyl-ethane, 1-[4-(2-hydroxyethoxy)-phenyl]-2-hydroxy-2-methyl-1-propan-1-one (IRGACURE® 2959); 2-hydroxy-1-[4-[4-(2-hydroxy-2-methyl-propionyl)-benzyl]-phenyl]-2-methyl-propan-1-one (IRGACURE® 127); 2-hydroxy-1-[4-[4-(2-hydroxy-2-methyl-propionyl)-phenoxy]-phenyl]-2-methyl-propan-1-one; dialkoxyacetophenones,  $\alpha$ -hydroxy- or  $\alpha$ -aminoacetophenones, e.g. (4-methylthiobenzoyl)-1-methyl-1-morpholinoethane (IRGACURE® 907), (4-morpholinobenzoyl)-1-benzyl-1-dimethylaminopropane (IRGACURE® 369), (4-morpholinobenzoyl)-1-(4-methylbenzyl)-1-dimethylaminopropane (IRGACURE® 379), (4-(2-hydroxyethyl)aminobenzoyl)-1-benzyl-1-dimethylaminopropane), (3,4-dimethoxybenzoyl)-1-benzyl-1-dimethylaminopropane; 4-aryl-1,3-dioxolanes, benzoin alkyl ethers and benzil ketals, e.g. dimethyl benzil ketal, phenylglyoxalic esters and derivatives thereof, e.g. methyl
- 10
- 15
- 20

$\alpha$ -oxo benzeneacetate, oxo-phenyl-acetic acid 2-(2-hydroxy-ethoxy)-ethyl ester, dimeric phenylglyoxalic esters, e.g. oxo-phenyl-acetic acid 1-methyl-2-[2-(2-oxo-2-phenyl-acetoxy)-propoxy]-ethyl ester (IRGACURE® 754); ketosulfones, e.g. ESACURE KIP 1001 M®; oxime-esters, e.g. 1,2-octanedione 1-[4-(phenylthio)phenyl]-2-(O-benzoyloxime) (IRGACURE®  
5 OXE01), ethanone 1-[9-ethyl-6-(2-methylbenzoyl)-9H-carbazol-3-yl]-1-(O-acetyloxime) (IRGACURE® OXE02), 9H-thioxanthene-2-carboxaldehyde 9-oxo-2-(O-acetyloxime), per-esters, e.g. benzophenone tetracarboxylic peresters as described for example in EP 126541, monoacyl phosphine oxides, e.g. (2,4,6-trimethylbenzoyl)diphenylphosphine oxide (DAROCUR® TPO), ethyl (2,4,6 trimethylbenzoyl phenyl) phosphinic acid ester; bisacyl-  
10 phosphine oxides, e.g. bis(2,6-dimethoxy-benzoyl)-(2,4,4-trimethyl-pentyl)phosphine oxide, bis(2,4,6-trimethylbenzoyl)-phenylphosphine oxide (IRGACURE® 819), bis(2,4,6-trimethylbenzoyl)-2,4-dipentoxyphenylphosphine oxide, trisacylphosphine oxides, halomethyltriazines, e.g. 2-[2-(4-methoxy-phenyl)-vinyl]-4,6-bis-trichloromethyl-[1,3,5]triazine, 2-(4-methoxy-phenyl)-4,6-bis-trichloromethyl-[1,3,5]triazine, 2-(3,4-dimethoxy-phenyl)-4,6-bis-trichlorome-  
15 thyl-[1,3,5]triazine, 2-methyl-4,6-bis-trichloromethyl-[1,3,5]triazine, hexaarylbisimidazole / coiniciators systems, e.g. ortho-chlorohexaphenyl-bisimidazole combined with 2-mercapto-benzthiazole, ferrocenium compounds, or titanocenes, e.g. bis(cyclopentadienyl)-bis(2,6-difluoro-3-pyrryl-phenyl)titanium (IRGACURE®784). Further, borate compounds can be used as coiniciators. As additional photoinitiators oligomeric compounds such as for example  
20 oligomeric alpha hydroxyl ketones e.g. 2-hydroxy-1-{1-[4-(2-hydroxy-2-methyl-propionyl)-phenyl]-1,3,3-trimethyl-indan-5-yl}-2-methyl-propan-1-one, ESACURE KIP® provided by Fratelli Lamberti, or oligomeric alpha amino ketones may be employed as well.

25 In one embodiment the photoinitiator is benzophenone (DAROCURE® BP), benzildimethylketal (IRGACURE® 651), 1-hydroxy-cyclohexyl-phenyl-ketone (IRGACUR 184), 1-[4-(2-hydroxyethoxy)-phenyl]-2-hydroxy-2-methyl-1-propan-1-one (IRGACURE® 2959), 2-hydroxy-1-{4-[4-(2-hydroxy-2-methyl-propionyl)-benzyl]-phenyl}-2-methyl-propan-1-one (IRGACURE®127), .(4-methylthio-  
30 benzoyl)-1-methyl-1-morpholinoethane (IRGACURE® 907), (4-morpholinobenzoyl)-1-benzyl-1-dimethylaminopropane (IRGACURE® 369), (4-morpholinobenzoyl)-1-(4-methylbenzyl)-1-dimethylaminopropane (IRGACURE® 379), (2,4,6-trimethylbenzoyl)diphenylphosphine oxide (DAROCUR® TPO), bis(2,6-dimethoxy-benzoyl)-(2,4,4-trimethyl-pentyl)phosphine oxide, bis(2,4,6-trimethylbenzoyl)-phenylphosphine oxide (IRGACURE® 819) and 2-hydroxy-1-{1-

[4-(2-hydroxy-2-methyl-propionyl)-phenyl]-1,3,3-trimethyl-indan-5-yl}-2-methyl-propan-1-one, (ESACURE ONE®), 1,2-octanedione, 1-[4-(phenylthio) phenyl]-2-(O-benzyloxime) (IRGACURE OXE01®). (claim 7)

5

The photoinitiator is preferably 1-hydroxy-cyclohexyl-phenyl-ketone (IRGACURE® 184), 2-hydroxy-1-{1-[4-(2-hydroxy-2-methyl-propionyl)-phenyl]-1,3,3-trimethyl-indan-5-yl}-2-methyl-propan-1-one, (ESACURE ONE®), bis(2,6-dimethoxy-benzoyl)-(2,4,4-trimethyl-pentyl)-phosphine oxide, bis(2,4,6-trimethylbenzoyl)-phenylphosphine oxide (IRGACURE® 819), 1-  
10 [4-(2-hydroxyethoxy)-phenyl]-2-hydroxy-2-methyl-1-propan-1-one (IRGACURE® 2959).

Most preferred are solid alpha hydroxyketones like e.g. IRGACURE® 184.

### Amounts

15 The photoinitiator is present in an amount of at least 30% by weight, more preferably at least 50% by weight and most preferably at least 65% by weight based on the total blend. (claim 2)

20 The alkoxyated liquid surface active agent is present in an amount between 1 and 70% by weight, more preferably between 1 and 40% by weight and most preferably between 1 and 20% by weight based on the total blend.

### Preparation

25 The inventive homogeneous liquid blend is obtainable by a process comprising the steps of

- heating at least 30% by weight of a photoinitiator to above its melting temperature
- adding 1-70% by weight of at least one alkoxyated liquid surface active agent of a liquid surface active agent and optionally adding a diluent
- cooling to room temperature to obtain a liquid blend.

30

### Advantages

The inventive free flowing liquid is suitable for aqueous and solvent based systems.

Higher active contents are obtainable as compared to previous solutions like the waterborne dispersions of photoinitiators described in EP386650 (Lamberti) and EP1230276B1 (Ciba) and in GB2014152 (Minesota Mining) In addition the inventive homogenous liquid blends show better storage stability, especially at elevated temperatures, as there are no  
5 heterogeneous photoinitiator particles which may irreversibly settle like in the case of a dispersion.

The invention further relates to a photopolymerizable composition comprising  
10 (A) at least one ethylenically unsaturated photopolymerizable compound and  
(B) at least a homogenous liquid blend of photoinitiator and surfactant as described above.

The unsaturated compounds may include one or more olefinic double bonds. They may be of low (monomeric) or high (oligomeric) molecular mass. Examples of monomers containing a double bond are alkyl or hydroxyalkyl acrylates or methacrylates, for example methyl, ethyl,  
15 butyl, 2-ethylhexyl or 2-hydroxyethyl acrylate, isobornyl acrylate, methyl methacrylate or ethyl methacrylate. Interesting also are resins which are modified with silicon or fluor, e.g. silicon acrylates. Other examples are acrylonitrile, acrylamide, methacrylamide, N-substituted (meth)acrylamides, vinyl esters such as vinyl acetate, vinyl ethers such as isobutyl vinyl ether, styrene, alkyl- and halostyrenes, N-vinylpyrrolidone, vinyl chloride or vinylidene  
20 chloride.

Examples of monomers containing two or more double bonds are the diacrylates of ethylene glycol, propylene glycol, neopentyl glycol, hexamethylene glycol or of bisphenol A, and 4,4'-bis(2-acryloyloxyethoxy)diphenylpropane, trimethylolpropane triacrylate, pentaerythritol  
25 triacrylate or tetraacrylate, vinyl acrylate, divinylbenzene, divinyl succinate, diallyl phthalate, triallyl phosphate, triallyl isocyanurate or tris(2-acryloylethyl) isocyanurate.

Examples of polyunsaturated compounds of relatively high molecular mass (oligomers) are acrylated epoxy resins, acrylated polyesters, polyesters containing vinyl ether or epoxy  
30 groups, and also acrylated polyurethanes and polyethers. Further examples of unsaturated oligomers are unsaturated polyester resins, which are usually prepared from maleic acid, phthalic acid and one or more diols and have molecular weights of from about 500 to 3000. In addition it is also possible to employ vinyl ether monomers and oligomers, and also

maleate-terminated oligomers with polyester, polyurethane, polyether, polyvinyl ether and epoxy main chains.

5 Particularly suitable examples are esters of ethylenically unsaturated carboxylic acids and polyols or polyepoxides, and polymers having ethylenically unsaturated groups in the chain or in side groups, for example unsaturated polyesters, polyamides and polyurethanes and copolymers thereof, alkyd resins, polybutadiene and butadiene copolymers, polyisoprene and isoprene copolymers, polymers and copolymers containing (meth)acrylic groups in side chains, and also mixtures of one or more such polymers.

10

Examples of unsaturated carboxylic acids are acrylic acid, methacrylic acid, crotonic acid, itaconic acid, cinnamic acid, and unsaturated fatty acids such as linolenic acid or oleic acid. Acrylic and methacrylic acid are preferred.

15 Suitable polyols are aromatic and, in particular, aliphatic and cycloaliphatic polyols.

Examples of aromatic polyols are hydroquinone, 4,4'-dihydroxydiphenyl, 2,2-di(4-hydroxyphenyl)propane, and also novolaks and resols. Examples of polyepoxides are those based on the abovementioned polyols, especially the aromatic polyols, and epichlorohydrin. Other suitable polyols are polymers and copolymers containing hydroxyl groups in the polymer  
20 chain or in side groups, examples being polyvinyl alcohol and copolymers thereof or polyhydroxyalkyl methacrylates or copolymers thereof. Further polyols which are suitable are oligoesters having hydroxyl end groups.

Examples of aliphatic and cycloaliphatic polyols are alkylene diols having preferably 2 to 12 C  
25 atoms, such as ethylene glycol, 1,2- or 1,3-propanediol, 1,2-, 1,3- or 1,4-butanediol, pentanediol, hexanediol, octanediol, dodecanediol, diethylene glycol, triethylene glycol, polyethylene glycols having molecular weights of preferably from 200 to 1500, 1,3-cyclopentanediol, 1,2-, 1,3- or 1,4-cyclohexanediol, 1,4-dihydroxymethylcyclohexane, glycerol, tris( $\beta$ -hydroxyethyl)amine, trimethylolethane, trimethylolpropane, pentaerythritol, dipentaerythritol  
30 and sorbitol.

The polyols may be partially or completely esterified with one carboxylic acid or with different unsaturated carboxylic acids, and in partial esters the free hydroxyl groups may be modified, for example etherified or esterified with other carboxylic acids.

Examples of esters are:

5 trimethylolpropane triacrylate, trimethylolethane triacrylate, trimethylolpropane trimeth-  
acrylate, trimethylolethane trimethacrylate, tetramethylene glycol dimethacrylate, triethylene  
glycol dimethacrylate, tetraethylene glycol diacrylate, pentaerythritol diacrylate, penta-  
erythritol triacrylate, pentaerythritol tetraacrylate, dipentaerythritol diacrylate, dipentaerythritol  
10 triacrylate, dipentaerythritol tetraacrylate, dipentaerythritol pentaacrylate, dipentaerythritol  
hexaacrylate, tripentaerythritol octaacrylate, pentaerythritol dimethacrylate, pentaerythritol  
trimethacrylate, dipentaerythritol dimethacrylate, dipentaerythritol tetramethacrylate,  
15 tripentaerythritol octamethacrylate, pentaerythritol diitaconate, dipentaerythritol tris-itaconate,  
dipentaerythritol pentaitaconate, dipentaerythritol hexaitaconate, ethylene glycol diacrylate,  
1,3-butanediol diacrylate, 1,3-butanediol dimethacrylate, 1,4-butanediol diitaconate, sorbitol  
triacrylate, sorbitol tetraacrylate, pentaerythritol-modified triacrylate, sorbitol tetra  
methacrylate, sorbitol pentaacrylate, sorbitol hexaacrylate, oligoester acrylates and  
20 methacrylates, glycerol diacrylate and triacrylate, 1,4-cyclohexane diacrylate, bisacrylates  
and bismethacrylates of polyethylene glycol with a molecular weight of from 200 to 1500, or  
mixtures thereof.

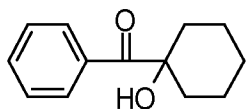
## 20 Use

The homogeneous liquid blend as defined above is used to produce solvent-free, solvent- or water-based surface coatings, printing inks, screen-printing inks, offset-printing inks, flexographic-printing inks, resist materials or image-recording material, especially in the production of corresponding aqueous systems.

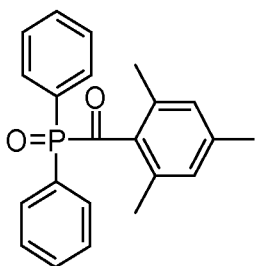
### Examples and application results

Liquid photoinitiator product forms of the following photoinitiators were prepared:

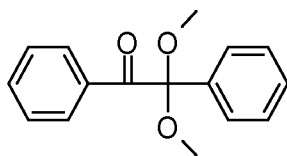
5 IRGACUR® 184



DAROCURE® TPO

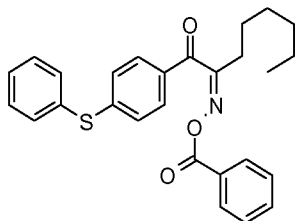


IRGACURE® 651



10

IRGACURE® OXE 01



ESACURE® ONE Difunctional Alpha Hydroxyketone, available from Lamberti

- 15 ESACURE KIP 150® Oligomeric polyfunctional alpha-hydroxyketone, available from Lamberti.

**Example 1** 75 g of 1-hydroxy-cyclohexyl-phenyl-ketone (IRGACURE 184) with a purity >90% is molten in a reactor at 60 °C. 10 g of Hedipin R/200 (PEG-20 Castor oil) is added, 15 g of tetraethylene glycol is added and the mixture is stirred at 60 °C until homogeneous. The

20

mixture is cooled slowly to room temperature. The product form is a clear, colorless to yellow liquid.

**Example 2** 75 g of 1-hydroxy-cyclohexyl-phenyl-ketone (IRGACURE 184) with a purity >90% is molten in a reactor at 60 °C. 10 g of Hedipin R/200 (PEG-20 Castor oil) is added, 15 g of poly(ethylene glycol) 350 monomethyl ether is added and the mixture is stirred at 60 °C until homogeneous. The mixture is cooled slowly to room temperature. The product form is a clear, colorless to yellow liquid.

**Example 3** 70 g of 1-hydroxy-cyclohexyl-phenyl-ketone (IRGACURE 184) with a purity >90% is molten in a reactor at 60 °C. 8 g of Tween 80 (poly(ethylene glycol) sorbitan monooleate) and 2 g of Span 80 (sorbitan monooleate) is added, 20 g poly(ethylene glycol) 350 monomethyl ether is added and the mixture is stirred at 60 °C until homogeneous. The mixture is cooled slowly to room temperature. The product form is a clear, colorless to yellow liquid.

**Example 4** 90 g of 1-hydroxy-cyclohexyl-phenyl-ketone (IRGACURE 184) with a purity >90% is molten in a reactor at 60 °C. 10 g of Brij 30 (tetraethylene glycol dodecyl ether) is added and the mixture is stirred at 60 °C until homogeneous. The mixture is cooled slowly to room temperature. The product form is a clear, colorless to yellow liquid.

**Example 5** 31 g of 2,4,6-trimethylbenzoyl-diphenyl-phosphineoxide (DAROCURE TPO) is mixed with 10 g of Hedipin R/200 (PEG-20 Castor oil) and 59 g of poly(ethylene glycol) 350 monomethyl ether. The mixture is stirred in a reactor at 90 °C until homogeneous. The mixture is cooled slowly to room temperature. The product form is a clear yellowish liquid.

**Example 6** 50 g of 1,2-octanedione, 1-[4-(phenylthio) phenyl]-,2-(O-benzyloxime) (IRGACURE OXE01) is mixed with 10 g of Hedipin R/200 (PEG-20 Castor oil) and 40 g of poly(ethylene glycol) 350 monomethyl ether. The mixture is stirred in a reactor at 60 °C until homogeneous. The mixture is cooled slowly to room temperature. The product form is a clear yellowish liquid.

**Example 7** 31 g of 2,2-Dimethoxy-1,2-diphenylethan-1-one (IRGACURE 651) is mixed with 10 g of Hedipin R/200 (PEG-20 Castor oil) and 59 g of poly(ethylene glycol) 350 monomethyl

ether. The mixture is stirred in a reactor at 70 °C until homogeneous. The mixture is cooled slowly to room temperature. The product form is a clear yellowish liquid.

### Example 8

- 5 70 g of 2-hydroxy-1-{1-[4-(2-hydroxy-2-methyl-propionyl)-phenyl]-1,3,3-trimethyl-indan-5-yl}-2-methyl-propan-1-one (Esacure ONE) is mixed with 8 g of Tween 80 (poly(ethylene glycol) sorbitan monooleate), 2 g of Span 80 (sorbitan monooleate) and 20 g of tetraethylene glycol. The mixture is heated to 115 °C and stirred until homogeneous. The mixture is cooled slowly to room temperature. The product form is a clear, colorless to yellow viscous liquid.

10

### Application results

In the following the photoinitiator refers to its inventive liquid blend delivery form, not to the active content.

- 15 1. UV-curing of clear Epoxy-Acrylate Coatings

A UV-curable paint is prepared using the following formulation:

% by weight	Component	Description	Supplier
89.0	Ebecryl 604	Unsaturated epoxy-acrylate approx. 80% in HDDA	Cytec Surface Specialties
10.0	SR 344	Polyethylene Glycol (400) Diacrylate	Sartomer
1.0	Ebecryl 350	Unsaturated silicone acrylate	Cytec Surface Specialties

- 20 Samples were prepared using above formulation and dissolving 2% by wt of a photoinitiator (delivery form) into the above paint. For that, the samples were stirred at a temperature of 50 °C for 60 min.

Samples were applied onto white pre-coated aluminium panels using a wire-wound draw down bar with WFT 60 µm. The samples were cured on an IST laboratory UV-curing equipment using one Hg bulb at 100 W/cm and a line speed of 10 m/min.

- 25 The extend of curing was followed by measurement of acrylate double bond conversion (DBC) against the resin itself as reference using AT-IR spectroscopy of the coating surface.

The acrylate peak at approx.  $1425\text{ cm}^{-1}$  was normalized using the CH-bond at approx.  $2900\text{ cm}^{-1}$ . In addition pendulum hardness according to Koenig was measured on all panels.

Curing Results:

	<b>ICU 184</b>	<b>Ex 1</b>	<b>Ex 2</b>	<b>Ex 3</b>	<b>Ex 4</b>
<b>active content (%)</b>	<b>100</b>	<b>75</b>	<b>75</b>	<b>70</b>	<b>90</b>
DBC [%]	79	74	73	70	76
PH [s]	133	128.8	128.3	127	127

5 These curing results demonstrate that the inventive liquid photoinitiator blends provide good curing performance in a typical 100% UV system, with double bond conversion and pendulum hardness similar to a state of the art photoinitiator ICU 184 which is not diluted by blend components.

Goal here is to show that the liquid blend gives a good performance in 100% system (advantage would be the liquid product form).

A further advantage is shown for the aqueous systems

## 2. UV-curing of clear Polyester-Acrylate Coatings

A UV-curable paint is prepared using the following formulation:

<b>% by weight</b>	<b>Component</b>	<b>Description</b>	<b>Supplier</b>
67.0	Ebecryl 830	hexafunctional polyester-acrylate	Cytec Surface Specialties
16.5	HDDA	Hexandiol Diacrylate, reactive diluent	
16.5	TMPTA	Trimethylolpropane Triacrylate, reactive diluent	

15

Samples were prepared using above formulation and dissolving 6% by wt of a photoinitiator preparation into the above paint. For that, the samples were stirred at a temperature of  $50\text{ }^{\circ}\text{C}$  for 60 min.

Samples were applied onto white pre-coated aluminium panels using a wire-wound draw down bar with WFT 60  $\mu\text{m}$ . The samples were cured on an IST laboratory UV-curing equipment using two Hg bulbs at 100 W/cm and a line speed of 10 m/min.

- 5 The extend of curing was followed by measurement of acrylate double bond conversion (DBC) against the resin itself as reference using AT-IR spectroscopy of the coating surface. The acrylate peak at approx.  $1425\text{ cm}^{-1}$  was normalized using the CH-bond at approx.  $2900\text{ cm}^{-1}$ . In addition pendulum hardness according to Koenig was measured on all panels.

#### 10 Curing Results

	<b>ICU 184</b>	<b>Ex 1</b>	<b>Ex 2</b>	<b>Ex 3</b>	<b>Ex 4</b>
<b>active content (%)</b>	<b>100</b>	<b>75</b>	<b>75</b>	<b>70</b>	<b>90</b>
DBC [%]	80	72	72	70	74
PH [s]	139	135	135	130	136

Again the curing results demonstrate that the liquid photoinitiator blends provide useful good curing performance despite the dilution with blend components.

#### 15

##### 3. Incorporation in w/b clear UV-curable formulation

The following water-based formulation was prepared (WB-Formulation 1):

<b>% by weight</b>	<b>Product</b>	<b>Description</b>	<b>Supplier</b>
99.3	Bayhydrol UV 2282	UV-curable Polyurethane dispersion in water	Bayer Material Science
0.30	Borchigel L 75 N	Polyurethane thickener, approx. 50% in water	OMG Borchers GmbH
0.40	Byk 347	Silicone surfactant for aqueous systems	Byk Additives and Instruments

- 20 0.4 g photoinitiator (delivery form) were placed into a 100 ml glass bottle. 40 g of above water-based formulation was added. The combination was stirred using a magnetic stirring bar for 5 min at approx. 500 rpm. Immediately after mixing, the coating samples were applied

over black pre-coated steel panel using a wire-wound draw down bar with wet film thickness of 120  $\mu\text{m}$ .

The films were dried for 30 min at 30 °C and then were cured on an IST laboratory UV-curing equipment using two Hg bulbs at 100 W/cm and a line speed of 10 m/min.

- 5 The films were evaluated visually for particles or smear traces of poorly incorporated photoinitiator.

Photoinitiator	Visual evaluation after application (wet film)	Visual evaluation after curing
Irgacure 184	Undissolved particles	Undissolved particles
Example 1	Good	Good
Example 3	Good	Good

- 10 These results show that the inventive photoinitiator blends can be easily incorporated into a waterborne UV-formulation without giving defects to the paint film. This is in contrast to the reference Irgacure 184, which due to the incompatibility does not give a useful waterborne UV-paint. Thus the inventive photoinitiator blends have a much broader applicability ranging from typical non-aqueous UV-paints up to aqueous UV-systems.

#### 4. Curing of w/b clear UV-curable formulation

- 15 The photoinitiator samples were tested in 2 different water-based clear coat formulations.

WB-Formulation 1: as described above

WB-Formulation 2:

% by weight	Product	Description	Supplier
98.1	Alberdingk LUX 399	UV-curable aliphatic polyurethane copolymer	Alberdingk Boley
0.5	Byk 024	Defoamer for aqueous systems	Byk Additives and Instruments
0.3	Byk 333	Silicone based surface additive	
0.3	Byk 348	Silicone surfactant for aqueous systems	
0.8	Acrysol RM 825	Non-ionic thickner for aqueous systems	Rohm and Haas

- 20 30 g Formulation was prepared containing 1% by weight photoinitiator (delivery form). The formulation was placed into a glass bottle and photoinitiator was added during stirring with toothed disk at 1500 rpm. The formulations were stirred for a total of 15 min. Samples were

applied over white pre-coated aluminium panel using a wire-wound draw down bar with wet film thickness of 120 µm.

Water was evaporated in a convection oven at 60 °C for 10 min. The film was cured on an IST laboratory UV-curing equipment using two Hg bulbs at 100 W/cm and a line speed of 10 m/min. After cooling, the pendulum hardness according to Koenig was measured.

For reference, a photoinitiator free sample of the formulations was treated in the same way.

Curing Results WB-Formulation 1:

Photoinitiator	Pendulum Hardness [s]
No photoinitiator	21
Example 1	106
Example 2	111

Curing Results WB-Formulation 2:

Photoinitiator	Pendulum Hardness [s]
No photoinitiator	Not determined, sticky surface
Example 4	112
Example 3	111

These results show that the inventive liquid photoinitiator blends in addition to conventional non-aqueous UV systems, also give good curing performance in waterborne UV-paints.

### Example 9

70 g of ESACURE KIP 150 is heated to 70 °C. 10 g of Hedipin R/200 (PEG-20 Castor oil) and 20 g of tetraethylene glycol are added and the mixture is stirred at 70 °C until homogeneous. The homogeneous mixture is cooled slowly to room temperature. The product form is a clear yellowish liquid.

### Application results

The application properties of example 9 were assessed in an overprint varnish (OPV). Example 9 was tested against Esacure KIP 150 in its pure form. Assessed were handling, incorporation, curing efficiency and yellowing (visual).

5

#### Testing system

OPV with aminoacrylate	Weight [g]
Ebecryl 605 (Bisphenol A epoxyacrylate diluted with 25% of TPGDA)	30.0
Ebecryl 7100 (amine modified acrylate)	10.0
Ebecryl 40 (propoxylated/ethoxylated pentaerythritol tetraacrylate)	5.0
OTA 480 (propoxylated glycerol triacrylate)	30.0
TPGDA (tripropylenglycol diacrylate)	24.0
Ebecryl 1360 (silicone hexaacrylate)	0.5
Dow Corning 57 (silicone additive)	0.5
$\Sigma$	100.0

#### Handling

Esacure KIP 150 had to be extracted with a hammer and was dissolved for 1 hour at 45°C in the OPV. By contrast, Example 9 had a viscosity similar to honey and could be directly poured into the OPV.

10

#### Curing efficiency

Both KIP 150 and Example 9 were tested at 3% active. Cure speeds were compared on cardboard and coilcoat.

15

Formulation	Substrate	%	m/min	KMnO <sub>4</sub> OD
Esacure KIP 150	Cardboard	3	<b>90 - 100</b>	0,309
Example 9		4,29	<b>100</b>	0,301
Esacure KIP 150	Coilcoat	3	<b>110</b>	0,373
Example 9		4,29	<b>120</b>	0,341

Table 1: Cold mirror IST, 1% O<sub>2</sub>, cardboard, bar #1 (6 µm), 80 W/cm, Hg lamp. DRR test

Table 1 shows that the reactivity of Example 9 is comparable to Esacure KIP 150 in the pure product form.

20

Visually, no difference in yellowing could be observed.

5 These application results show that the liquid product form provides clear processing advantages (handling and incorporation), without compromising curing efficiency.

**Claims**

1. A homogenous liquid blend which is liquid at room temperature comprising at least 30% by weight with regard to the whole blend of a solid photoinitiator of the radical type and 1-70% by weight of at least one alkoxyated liquid surface active agent and optionally an organic diluent.
- 5
2. A homogeneous liquid blend according to claim 1 wherein the photoinitiator concentration is at least at least 50% by weight and most preferable at least 65% by weight.
- 10
3. A homogenous liquid blend according to claim 1 wherein the surface active agent is selected from alkoxyated compounds based on adducts with ethylene oxide and/or propylene oxide.
- 15
4. A homogenous liquid blend according to claim 1 wherein the surface active agent is selected from alcohol alkoxyates (glycolethers), fatty acid alkoxyates, fatty amine alkoxyates, or fatty amide alkoxyates, alkoxyates of phenols or alkylsubstituted phenols, block copolymers or random copolymers of ethylene oxide and propylene oxide, alkoxyates of alkine-alcohols or alkine-diols, alkoxyates of castor oil, alkoxyates of sorbitan fatty acid esters and mixtures thereof.
- 20
5. A homogenous liquid blend according to claim 1 wherein the diluent is present.
- 25
6. A homogenous liquid blend according to claim 1 wherein the photoinitiator is selected from benzophenones, aromatic  $\alpha$ -hydroxyketones, benzilketals, aromatic  $\alpha$ -aminoketones, phenylglyoxalic acid esters, mono-acylphosphinoxides, bis-acylphosphinoxides, tris-acylphosphinoxides and oximesters derived from aromatic ketones.
- 30

7. A homogenous liquid blend according to claim 1 wherein the photoinitiator is selected from benzophenone (DAROCURE® BP), benzildimethylketal (IRGACURE® 651), 1-hydroxy-cyclohexyl-phenyl-ketone (IRGACUR 184), 1-[4-(2-hydroxyethoxy)-phenyl]-2-hydroxy-2-methyl-1-propan-1-one (IRGACURE® 2959), 2-hydroxy-1-{4-[4-(2-hydroxy-2-methyl-propionyl)-benzyl]-phenyl}-2-methyl-propan-1-one (IRGACURE®127), .(4-methylthio-benzoyl)-1-methyl-1-morpholinoethane (IRGACURE® 907), (4-morpholinobenzoyl)-1-benzyl-1-dimethylaminopropane (IRGACURE® 369), (4-morpholinobenzoyl)-1-(4-methylbenzyl)-1-dimethylaminopropane (IRGACURE® 379), (2,4,6-trimethylbenzoyl)diphenylphosphine oxide (DAROCUR® TPO), bis(2,6-dimethoxy-benzoyl)-(2,4,4-trimethyl-pentyl)phosphine oxide, bis(2,4,6-trimethylbenzoyl)-phenylphosphine oxide (IRGACURE® 819) and 2-hydroxy-1-{1-[4-(2-hydroxy-2-methyl-propionyl)-phenyl]-1,3,3-trimethyl-indan-5-yl}-2-methyl-propan-1-one, (ESACURE ONE®), 1,2-octanedione, 1-[4-(phenylthio) phenyl]-2-(O-benzyloxime) (IRGACURE OXE01®).

15

8. A homogenous liquid blend according to claim 7 wherein the photoinitiator is selected from 1-hydroxy-cyclohexyl-phenyl-ketone (IRGACURE 184), 2-hydroxy-1-{1-[4-(2-hydroxy-2-methylpropionyl)-phenyl]-1,3,3-trimethyl-indan-5-yl}-2-methyl-propan-1-one, (ESACURE ONE), bis(2,6-dimethoxy-benzoyl)-(2,4,4-trimethyl-pentyl)phosphine oxide, bis(2,4,6-trimethylbenzoyl)-phenylphosphine oxide (IRGACURE® 819), 1-[4-(2-hydroxyethoxy)-phenyl]-2-hydroxy-2-methyl-1-propan-1-one (IRGACURE 2959).

25

9. A process for the preparation of a homogenous liquid blend as defined in claim 1 comprising the steps of

- heating at least 30% by weight of a photoinitiator to a temperature above its melting point
- adding 1-70% by weight of at least one alkoxyated liquid surface active agent and optionally adding a diluent
- cooling to room temperature to obtain a homogeneous liquid blend.

30

10. A photopolymerisable composition comprising

(A) at least one ethylenically unsaturated photopolymerisable compound and

(B) at least a homogenous liquid blend according to claim 1.

5 **11.** The use of a homogeneous liquid blend according to claim 1 in the production of solvent-free, solvent- and water-based surface coatings, printing inks, screen-printing inks, offset-printing inks, flexographic-printing inks, resist materials or image-recording material.

**INTERNATIONAL SEARCH REPORT**

International application No  
PCT/EP2010/050003

**A. CLASSIFICATION OF SUBJECT MATTER**  
INV. C08F2/50 C08F2/48

According to International Patent Classification (IPC) or to both national classification and IPC

**B. FIELDS SEARCHED**

Minimum documentation searched (classification system followed by classification symbols)  
C08F

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal, WPI Data

**C. DOCUMENTS CONSIDERED TO BE RELEVANT**

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	GB 2 014 152 A (MINNESOTA MINING & MFG) 22 August 1979 (1979-08-22) cited in the application page 5; example 14; table I	1-8, 10-11
A	US 4 965 294 A (OHNGEMACH JOERG [DE] ET AL) 23 October 1990 (1990-10-23) the whole document	1-11

Further documents are listed in the continuation of Box C.

See patent family annex.

\* Special categories of cited documents :

- \*A\* document defining the general state of the art which is not considered to be of particular relevance
- \*E\* earlier document but published on or after the international filing date
- \*L\* document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
- \*O\* document referring to an oral disclosure, use, exhibition or other means
- \*P\* document published prior to the international filing date but later than the priority date claimed

- \*T\* later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
- \*X\* document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
- \*Y\* document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.
- \*&\* document member of the same patent family

Date of the actual completion of the international search

22 March 2010

Date of mailing of the international search report

06/04/2010

Name and mailing address of the ISA/

European Patent Office, P.B. 5818 Patentlaan 2  
NL - 2280 HV Rijswijk  
Tel. (+31-70) 340-2040,  
Fax: (+31-70) 340-3016

Authorized officer

Le Masson, Nicolas

# INTERNATIONAL SEARCH REPORT

Information on patent family members

International application No

PCT/EP2010/050003

Patent document cited in search report	Publication date	Patent family member(s)	Publication date	
GB 2014152	A	22-08-1979	AU 526849 B2	03-02-1983
			AU 4402479 A	16-08-1979
			BE 874005 A1	07-08-1979
			CA 1121093 A1	30-03-1982
			DE 2904450 A1	09-08-1979
			FR 2416922 A1	07-09-1979
			JP 54117600 A	12-09-1979
			US 4186108 A	29-01-1980
			ZA 7900527 A	30-07-1980
US 4965294	A	23-10-1990	AU 616008 B2	17-10-1991
			AU 3408389 A	09-11-1989
			DE 3815622 A1	16-11-1989
			EP 0341534 A2	15-11-1989
			JP 2022304 A	25-01-1990